How to ensure a successful tablet coating formulation development by using polyvinyl alcohol and HPMC in combination with calcium carbonate as a titanium dioxide alternative



Anja-Nadine Knüttel, Almut von der Brelie, Thomas Kipping, Alessandra Ambruosi

Purpose

- Recent ban of titanium dioxide (TiO₂; E171) for the use in food and nutritional supplements (European Commission). The regulation includes a commitment to review the necessity of titanium dioxide as a white colorant in medicinal products¹.
- It is challenging to find a suitable alternative for TiO₂ because of its high opacity.
- As an approved white drug colorant in the EU, calcium carbonate is of interest as a potential alternative to TiO₂ in the coating process.

Objectives

- Evaluate a suitable alternative pigment to TiO₂ for tablet coating.
- Deep assessment of calcium carbonate in terms of its opacifying potential as well as the impact on required process conditions.
- Recommended formulations with a comparable covering effect need to be identified.

Methods

Colored tablet cores were manufactured using a rotary tablet press (FETTE 1200i) at 74 rpm; compression force: 10 kN; height: 5.0 mm; tablet diameter: 11 mm; targeted tablet weight: 500 mg. A simplified core formulation with red iron oxide as pigment was used: 98.0% mannitol +0.5% iron oxide +1.5% magnesium stearate. The coating with Parteck® COAT EMPROVE® ESSENTIAL (Polyvinyl alcohol, MilliporeSigma) was performed using a rotating drum coater type LDCS from Vector Freund Corporation. To identify a relevant coloring of the core tablets, different concentrations of iron oxide were tested and determined by color measurements. The ratio of mannitol was adapted in respect of the amount of iron oxide to simulate a suitable coloring.



Figure 1.Pre-mixtures of mannitol with different amounts of iron oxide.

Color measurements were performed by using Konica Minolta Spectrophotometer CM-700d. At a level of 0.5% iron oxide a suitable coloration of tablet cores was achieved to simulate potential colored drug substances and to evaluate the covering properties of calcium carbonate in comparison to titanium dioxide.

Results

Based on the results it was found that CaCO₃ represents a viable alternative to TiO₂ by increasing the solid content in the coating dispersion without any negative effect on the viscosity or feasibility of the coating process in general. Although the effectiveness and efficiency to TiO₂ is not directly comparable, the covering effect of CaCO₃ can be enhanced by adapting the solid content of CaCO₃.

Ingredients	Solid content [%]	Sample/ Batch	Density [g/ml]	Dyn. Visc. [mPas · s]
Parteck® COAT (PVA)	10	Basis Coating solution	1.0135 1.0194	38.28 38.06
PVA + 30%* CaCO ₃ +	15	FC-002-22	1.0521 1.0524	15.24 13.91
PVA + 40%* CaCO ₃ +	20	FC-003-22	1.0940 1.0978	20.52 20.06
PVA + 40%* TiO ₂ +	20	FC-004-22	1.0984 1.0986	33.78 33.53
HPMC + 30%* CaCO ₃ +	15	FC-010-22	1.0515 1.0538	60.94 62.48
HPMC + 40%* CaCO ₃ +	20	FC-011-22	1.0710 1.0743	79.25 80.24

Table 1.

Characterization of different coating solutions at 25 °C; *In relation to the total solid content in the coating dispersion; ingredients include 10% talcum and 20% plasticizer; the amount of polymer was adapted to the individual formulation.

Coating liquid

It was observed that the addition of pigments can affect the viscosity of the spraying liquid.

Calcium carbonate has low influence on viscosity in comparison to basic coating solution whereas for titanium dioxide an increase of viscosity is observed. This effect is even more pronounced when using HPMC based polymers (see Table 1).

Film coated tablets

Trials with calcium carbonate

- Constant & homogenous coating process was achieved.
- Higher solid content and weight gain lead to a better covering of colored core tablets.
- Higher coating thickness is achieved at same process conditions compared to TiO₂ (see Figure 2).

Trials with titanium dioxide

- Requires higher speed and more time for homogenization of coating suspension with 30% w/w solid content.
- Higher solid content leads to material settling in coating pipe and spray gun blocking tendency.
- Lower coating thickness (see Figure 2).

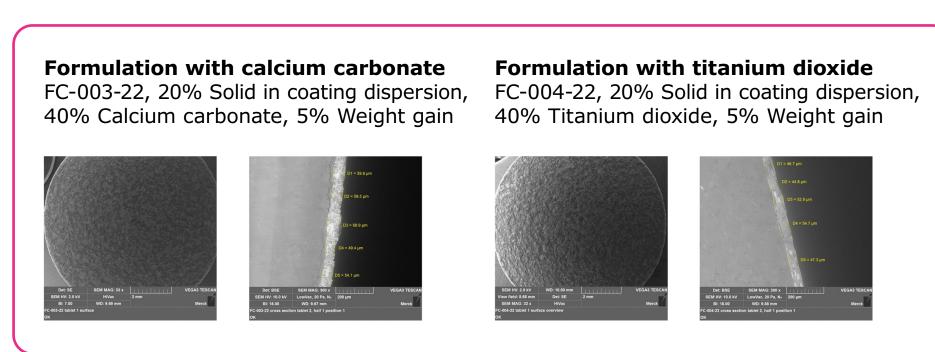


Figure 2. Individual SEM images of tablet surfaces (left) and cut tablet cores (right).

Calcium carbonate Raw tablets T22107 FC-001-22 98% Mannitol 9% Solid in 0.5% Irone oxide Coating Suspension 30% CaCO₃ 1.5% magnesium 3% Weight gain stearate FC-002-22 15% Solid in Coating Suspension 30% CaCO₃ 5% Weight gain Titanium dioxide FC-004-22 FC-003-22 20% Solid in 20% Solid in Coating Suspension Coating Suspension 40% CaCO₃ 40% TiO₂ 5% Weight gain 5% Weight gain FC-005-22 40% CaCO₃ 8% Weight gain FC-006-22 20% Solid in Coating Suspension 40% CaCO₃ ~20% Weight gain

Figure 3.Visual images of tablet surfaces coated with PVA-based Parteck® COAT Excipient and calcium carbonate vs. titanium dioxide.

Sample	Nr.	Starting conditions		Sample	NI	Starting conditions			
		L*	a*	b*	Sample	Nr.	L*	a*	b*
Mannitol [10.6 kN]	1 2 3 4 5		0.04 0.01 0.02	0.38 0.37 0.33 0.33 0.35		1	76.60	14.12	7.70
		99.36 99.37 99.44 99.47 99.46			Tablet	2	75.80	14.68	8.00
					cores	3	72.89	16.50	9.19
					COICS	4	76.23	14.54	8.01
					-	5	73.16	16.57	9.28
						1	73.66	16.62	11.54
					FC-001-22	2	69.74	18.73	12.24
Mannitol + 0.25% iron oxide	1	86.93 85.99 86.72 87.60 87.40	8.20 8.31 7.97 7.59 7.63	3.78 3.65 3.55 3.29 3.39	(CaCO ₃)	3	72.49	17.27	11.75
					(Caco ₃)	4	72.80	17.26	12.07
	2					5	73.05	16.95	11.55
	3					1	73.71	14.74	8.77
[10.6 kN]	4				FC-002-22	2	75.70	14.40	9.26
[1010 KH]	5				(CaCO ₃)	3	74.61	14.19	8.54
					(64663)	4	75.79	14.62	9.58
			10.83 11.35 11.15 10.71 11.86	4.79 5.20 5.43 5.03 5.62		5	74.35	14.33	8.54
Mannitol	1	80.89				1	80.03	9.89	6.12
+ 0.5% iron oxide [10.6 kN]	2 3 4 5	79.82 80.65 81.40 79.11			FC-003-22	2	79.32	10.03	5.90
					$(CaCO_3)$	3	80.40	9.63	6.06
					(3)	4	79.22	9.85	5.78
						5	79.43	9.33	5.46
Mannitol + 0.75% iron oxide [10.6 kN]	1 79.57 2 81.11 3 80.50 4 79.19 5 79.05	81.11	11.67 10.69	5.72 5.01		1	95.03	0.04	-0.98
					FC-004-22 (CaCO ₃)	2	94.43	0.10	-1.12
						3	94.10	0.27	-1.09
						4	94.60	0.15	-1.05
		10.94	4.85		5	94.70	0.02	-1.05	
				5.52 5.59		1	81.78	7.64	5.94
		/9.05			FC-005-22	2	80.81	8.48	5.87
					$(CaCO_3)$	3	81.01	8.16	5.87
			4 = 0 =	7.56 6.65 7.70 6.14 7.28	, 3,	4	79.52	8.79	5.62
Mannitol	1	72.77	1 14.58 1 15.16 1 13.13			5	81.02	8.64	6.18
+ 1.0% iron oxide [10.6 kN]	3 72. 4 75.	72.61				1	84.11	4.74	6.14
		72.61 75.34 73.93			FC-006-22	2	84.51	4.21	6.38
					$(CaCO_3)$	3	83.76	4.51	5.98
						4	83.59	5.03	5.90
						5	84.74	4.36	6.35

Table 2.Color measurements to identify the amount of iron oxide for core tablets (n=5).

Table 3.Color measurements of the film-coated tablets (n=5).

Conclusions

- The replacement of titanium dioxide by calcium carbonate has an interesting potential for future coating applications. In combination with polyvinyl alcohol low viscosities of the spraying liquid can be maintained even at high solid contents which can enable a high process efficiency
- This effect can potentially balance the required weight gains required to achieve the targeted covering effect of the core tablet surface.
- It could be demonstrated that iron oxide can be integrated in core-tablets to simulate stronger colored drug substances. Color measurements confirm the tendency that with higher calcium carbonate content the red coloring (parameter a) is further reduced indicating a stronger covering effect caused by higher opacity which is also confirmed by increasing L values (see Figure 4).

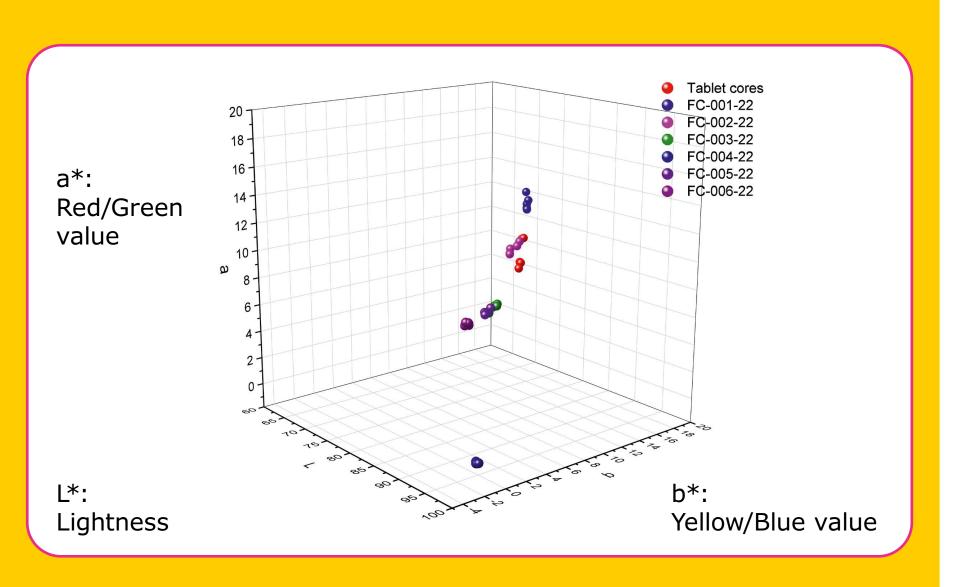


Figure 4. L*a*b color space distribution of coated tablets.

References

1. Official Journal of the European Union L 11/1 (18 Jan 2022) Commission Regulation (EU) 2022/63 of 14 January 2022.

Contact details: thomas.kipping@milliporesigma.com

We provide information and advice to our customers on application technologies and regulatory matters to the best of our knowledge and ability, but without obligation or liability. Existing laws and regulations are to be observed in all cases by our customers. This also applies in respect to any rights of third parties. Our information and advice do not relieve our customers of their own responsibility for checking the suitability of our products for the envisaged purpose.

SAFC

Pharma & Biopharma Raw Material Solutions

The Life Science business of Merck operates as MilliporeSigma in the U.S. and Canada.